

(LJ 4270) AUGUST 2016 Sub. Code: 4270

FOURTH YEAR B.PHARM. EXAMINATION PAPER IV – MODERN METHODS OF PHARMACEUTICAL ANALYSIS

Q.P. Code: 564270

Time: Three hours Maximum: 100 Marks

I. Essay: $(2 \times 20 = 40)$

- 1. With a neat diagram, explain the principle, different vibrational modes of molecules, instrumentation and applications of Infrared spectrophotomete
- 2. a) Explain the principle and instrumentation of flame emission spectroscopy.
 - b) Describe the principle and different types of conductometric titrations with examples.

II. Short notes: $(8 \times 5 = 40)$

- 1. Explain the various factors that affect the fluorescent intensity.
- 2. Describe the working principle and instrumentation of nephelomete
- 3. Explain the different type of electronic transitions involved in UV spectroscopy with examples.
- 4. Describe the construction and working of dropping mercury electrode with a diagram.
- 5. Describe briefly the various parameters involved in the calibration and validation of analytical instruments.
- 6. Explain the principle, different type of ion exchange resins and mechanism involved in ion exchange chromatography.
- 7. Explain the principle and applications of X-ray diffraction technique.
- 8. Explain the various methods of detection of compounds in thin layer chromatography.

III. Short answers: $(10 \times 2 = 20)$

- 1. Define chromophore and hypsochromic shift.
- 2. Define retention time and retention volume.
- 3. What is ultracentrifugation?
- 4. List at least four different type ions produced in mass spectra.
- 5. What is HETP? How it is related to number of theoretical plates and length of the column?
- 6. Mention two examples each of reference and indicator electrode.
- 7. Define finger print region.
- 8. Write any two differences between NMR and ES
- 9. Define tailing facto
- 10. Write any two limitations of GLC.
