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Course Content

(Based on Competency Table published by Medical Council of India.Students/Teachers are directed to a published on MCI Website for details)

Subject: Biochemistry
Year: First MBBS

Competency No.	Topics & Subtopics
1	Basic Biochemistry
1.1 P.2	Molecular and functional organization of cell and its sul
Describe the molecular and functional organization of a cell and its subcellular components.	
2 2	Enzymes
2.1 Explain fundamental concepts of enzyme, isoenzyme, alloenzyme, coenzyme & co-factors. Enumerate the main classes of IUBMB nomenclature.	Biochemical nature of enzyme, isoenzyme, alloenzyme, IUBMB enzyme classification
	Estimation of SGOT (AST)& SGPT (ALT) with its normal r
2.2 Observe the estimation of SGOT & SGPT	
2.3 Describe and explain the basic principles of enzyme activity	Mechanism of enzyme action, factors affecting enzyme kinetics with special reference to V _{max} & k _m .
2.4	Enzyme inhibition. Various inhibitors as drugs and poiso





Competency No.	Topics & Subtopics
Describe and discuss enzyme inhibitors as poisons and drugs and as therapeutic enzymes	•
2.5 Describe and discuss the clinical utility of various serum enzymes as markers of pathological conditions.	Diagnostic and therapeutic importance of various serun
2.6 Discuss use of enzymes in laboratory investigations (Enzymebased assays)	Analytical uses of Enzymes in laboratory investigations
2.7 Interpret laboratory results of enzyme activities & describe the clinical utility of various enzymes as markers of pathological conditions.	Interpret various serum enzymes of liver & biliary tract, in various disorders
3	Chemistry & Metabolism of Carbohydrates
3.1 Discuss and differentiate monosaccharides, di-saccharides and polysaccharides giving examples of main carbohydrates as energy fuel, structural element and storage in the human body	Classification of carbohydrates with examples and function examples as energy fuel, glycosides and its therapeutic examples and importance, polysaccharides with examp glycogen, structural elements like glycosaminoglycan's starch, glycemic index, and dietary fiber. Clinical import



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Digestion & absorption, transport and storage of carbo

3.2 Describe the processes involved in digestion and assimilation of carbohydrates and storage. 3.3	sucrase deficiency disorders
Competency No.	Topics & Subtopics
Describe and discuss the digestion and assimilation of carbohydrates from food.	
3.4 Define and differentiate the pathways of carbohydrate metabolism(glycolysis, gluconeogenesis, glycogen metabolism, HMP shunt). 3.5 Describe and discuss the regulation, functions and integration of carbohydrate along with associated diseases/disorders.	Pathway, energetics, regulation & clinical diseases / dis Rappaport Leubering cycle, Gluconeogenesis, Glycogen pathway, Uronic acid pathway, Galactose & Fructose m
3.6 Describe and discuss the concept of TCA cycle as a amphibolic pathway and its regulation.	TCA cycle Pathway, energetics, regulation & its concepts
3.7 To be clubbed with 3.4 & 3.6 Describe the common poisons that inhibit crucial enzymes of carbohydrate metabolism (eg: fluoride, arsenate)	Common poisons that inhibit crucial enzymes of carboh lodoacetate, fluoride & arsenite as poisons that inhibit Fluoroacetate, arsenite & malonate as poisons that inhi



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3.8 & 3.10

3.8: Discuss and interpret laboratory results of analytes associated with metabolism of carbohydrates. (to be clubbed with comp no 11.17- Diabetes Mellitus)

3.10

Interpret the results of blood glucose levels and other Laboratory investigations related to disorders of carbohydrate metabolism. Interpretation of the results of blood glucose, Glycated guidelines in Diabetes mellitus including gestational dia investigation like urinary glucose, urinary ketone bodies

Interpretation of the results of blood & urinary galactos

Interpretation of blood G6PD levels

Competency No.	Topics & Subtopics
3.9 Discuss the mechanism and significance of blood glucose regulation in health and disease.	Regulation of blood glucose in fed and fasting state in n diabetes mellitus.
4 200	Chemistry & Metabolism of Lipids
4.1 Describe and discuss main classes of lipids (Essential/nonessential fatty acids, cholesterol and hormonal steroids, triglycerides, major phospholipids and sphingolipids) relevant to human system and their major functions.	Definition & classification of lipids including classification nomenclature, numbering, functions & biological importacids, cholesterol, hormonal steroids, triglycerides, masphingolipids



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Topics & Subtopics

Metabolism of various lipoproteins and hyperlipoprotein

hypolipoproteinemiasabetalipoproteinemias & Tangiers

Classification structure and functions of lipoproteins- (7

Metabolic interrelationship between various lipoprotei

transport of cholesterol and reverse cholesterol transport

clubbed with 4.3)

Metabolism of fatty acids (β-oxidation of even and odd energetics and disorders associated with oxidation of fa
ketone bodies, its significance, regulation and associate
In brief de novo fatty acid biosynthesis- site & organs, p product formed & regulatory steps.
Biosynthesis of triacylglycerol and fate of triacylglycerol its significance and regulation, Metabolic role of adipos transport and storage like fatty liver.
In brief Cholesterol biosynthesis- site & organs, precurs & regulatory step, metabolic fate & excretion
()

disorders.

4.3

Explain the regulation of lipoprotein metabolism & associated

Describe the structure and functions of lipoproteins, their

functions, interrelations & relations with atherosclerosis



4.5 & 4.7 Interpret laboratory results of analytes associated with metabolism of lipids	Various lipid profile tests with their biological reference results in various disorders like hyper/hypolipoproteine syndrome, disorders of thyroid etc.
4.6 Describe the therapeutic uses of prostaglandins and inhibitors of eicosanoid synthesis.	Various eicosanoid classes (prostaglandins, leukotriene functions. Key features of synthesis of eicosanoids and therapeutic uses of prostaglandins
4.7 Interpret laboratory results of analytes associated with metabolism of lipids.	Same as 4.5
5 5.1 Describe and discuss structural organization of proteins.	Chemistry and Metabolism of Proteins General nature of amino acid, classification and imports examples, peptide bond formation, biologically imports protein structure including disulfide & weak bonds with significance.
Competency No.	Topics & Subtopics
5.2 Describe and discuss functions of proteins and structurefunction relationships in relevant areas e.g. hemoglobin and selected hemoglobinopathies	Definition, various classifications with examples and fur proteins, structure - function relationship of proteins lik hemoglobin



5.3	Digestion, absorption and transport of dietary proteins
Describe the digestion and absorption of dietary proteins.	Hartnup disease, cystinuria & glycinuria.
5.4 Describe common disorders associated with protein metabolism.	Role of transamination & deamination reactions in meta formation of ammonia with their clinical significance. Transport of ammonia, pathway of urea cycle, its significance disorders associated with urea cycle. Metabolic pathways for Glycine, Phenylalanine & Tyrozacids (Methionine, Cysteine & Cystine) and branch chai & Leucine), their role in biosynthesis of variety of specimetabolic disorders
5.5 Interpret laboratory results of analytesassociated with metabolism of proteins.	For Tryptophan- Only important biomolecules formed & Interpret laboratory results of protein metabolism for e Levels of various metabolites in blood or urine in metab disorders, Phenylketonuria, Tyrosinemia, Alkaptonuria, cystinuria & homocystinuria
6 M	Metabolism and Homeostasis
6.1 Discuss the metabolic processes that take place in specific organs in the body in the fed and fasting states.	Integration of carbohydrate, protein and lipid metabolic level with its significance, Metabolic processes with role and starvation states.
Competency No.	Topics & Subtopics





6.2 Describe and discuss the metabolic processes in which nucleotides are involved.	Important steps in de novo biosynthesis of purine and pregulation, enzymes of the nucleotide biosynthesis that salvage pathway for the synthesis of purine nucleotides of purine and pyrimidine nucleotides.
6.3 Describe the common disorders associated with nucleotide metabolism. 6.4 Discuss the laboratory results of analytes associated with gout & Lesch-Nyhan syndrome.	Disorder of nucleotide metabolism like gout, Lesch-Nyh diagnostic tests & biochemical mechanism of nutritional Lab results of analytes related with gout & Lesch-Nyhar blood & urine and presence of urate crystals in synovial blood
6.5 Describe the biochemical role of vitamins in the body and explain the manifestations of their deficiency	Sources, biochemical functions, daily requirement and a soluble vitamins (Vitamin A, D, E & K). Sources, biochemical functions and deficiency manifest (Thiamine, Riboflavin, Niacin, Pantothenic acid, Pyridox and vitamin C)
6.6 Describe the biochemical processes involved in generation of energy in cells.	Electron transport chain, mechanism of oxidative phosp theory), substrate level phosphorylation, Uncouplers & chain, shuttle systems for transport of extra-mitochono
6.7 Describe the processes involved in maintenance of normal pH, water & electrolyte balance of body fluids and the derangements associated with these.	Acids, bases and buffers, mechanism of action of buffer normal pH of body fluids. Role of blood buffers, respirar of blood pH. Disorders associated with blood pH (acidosis and alkalo mechanisms, anion gap & its clinical importance.



Competency No.	Topics & Subtopics
Jes L	Total body water and its compartmental distribution, values potassium and chloride, their distribution and clinical collevel alterations, maintenance of normal water and electrolyte imbalance.
6.8 Discuss and interpret results of Arterial Blood Gas (ABG) analysis in various disorders.	Interpretation of results of arterial blood gas (ABG) ana
6.9 Describe the functions of various minerals in the body, their metabolism and homeostasis.	Dietary food sources, daily requirement, biochemical fu homeostasis of: Calcium, phosphorus & magnesium, tra iodine, iron, manganese, selenium & zinc)
6.10 Enumerate and describe the disorders associated with mineral metabolism.	Clinical conditions related to plasma level alterations of magnesium Trace elements (copper, fluoride, iodine, ire
6.11 Describe the functions of heme in the body and describe the processes involved in its metabolism and describe porphyrin metabolism	Structure and functions of hemoglobin, role of 2,3-bisp binding and delivery, biosynthesis of heme (iron contain functions in the body, disorders of heme biosynthesis (catabolism of heme, various types of jaundice



6.12 Describe the major types of hemoglobin and its derivatives	Types of normal human hemoglobin, types of normal a hemoglobin, various hemoglobinopathies: Sickle cell a
found in the body and their physiological/ pathological relevance.	
	1
Competency No.	OTT Toules 9 Subtracion
	Topics & Subtopics
6.13	
Describe the functions of the kidney, liver, thyroid and adrenal glands.	1. Functions of liver, disorders& liver function tests
6.14	2. Functions of kidney, disorders& kidney function test
Describe the tests that are commonly done in clinical practice	3. Functions of Thyroid, disorders& thyroid function tes
to assess the functions of these organs (kidney, liver, thyroid and adrenal glands).	4. Functions of Adrenals , disorders& Adrenal function
6.15	
Describe the abnormalities of kidney, liver, thyroid and adrenal glands.	
7	Molecular Biology
7.1	Structure and functions of nucleotides, biologically imp
Describe the structure and functions of DNA and RNA and	importance, major types of synthetic analogs of nucleo
outline the cell cycle	clinical significance, structure and functions of DNA an



7.2

Describe the processes involved in replication & repair of DNA

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Replication of DNA in Eukaryotes, inhibitors of DNA rep

repair systems of DNA

and the transcription & translation mechanisms.	Transcription in Eukaryotes and posttranscriptional mo transcription & its significance
	Genetic code and wobble hypothesis, Translation in Europrotein folding and posttranslational modifications
7.3	Causes and types of genetic mutations with examples.
	Regulation of Eukaryotic gene expression
Te.	
Competency No.	Tourise & Subbanies
Lillo	Topics & Subtopics
Describe gene mutations and basic mechanism of regulation of gene expression	
7.4 Describe applications of molecular technologies like Recombinant DNA technology, PCR in the diagnosis and treatment of diseases with genetic basis.	Recombinant DNA technology, restriction endonuclease recombinant DNA and its applications in medicine, DNA southern blotting, northern blotting & western blotting reaction and its application in medical diagnosis and tre
7.5	Mechanisms of biotransformation of xenobiotics & asso
Describe the role of xenobiotics in disease	
7.6 Describe the anti-oxidant defense systems in the body.	Enzymatic and non-enzymatic antioxidant defense syst





Free radical, biological sources of reactive oxygen speci oxidative stress, roll of oxidative stress in cancer, diabe
CO,
Nutrition
Importance of carbohydrates, lipids, proteins & vitamin types of dietary fibers and their importance in the diet.
Protein energy malnutrition, Kwashiorkor and Marasmi
T
Topics & Subtopics
Balanced diet in adult, in childhood and in pregnancy for diabetes mellitus & coronary heart disease
Causes, effects and health risk associated with overweigh





8.5 Summarize the nutritional importance of commonly used items of food including fruits and vegetables (macro-molecules & its importance)	Nutritional importance of commonly used items of food fish, fruits and vegetables and their normal dietary requ
9	Extracellular Matrix
9.1 List the functions and components of the extracellular matrix (ECM).	Types & functions of the extracellular matrix (ECM), Co proteoglycans, glycoproteins & major proteins of ECM
9.2 Discuss the involvement of ECM components in health and disease.	Disorders associated with components of ECM like Oste Syndrome , Mucopolysaccharidoses, Scurvy & Menkes
9.3 Describe protein targeting & sorting along with its associated disorders(It is non-core: N)	Types of protein targeting and sorting, disorders due to targeting signals and defects in peroxisomal matrix pro
10	Oncogenesis and Immunity
Competency No.	Topics & Subtopics
10.1 Describe the cancer initiation, promotion oncogenes & oncogene activation. Also focus on p53 & apoptosis	Characteristics of cancer cell, molecular basis of cancer carcinogens and initiator, promoter of carcinogens, one tumor suppressor genes (retinoblastoma, RB and p53), physiologic and pathologic conditions.





Biochemical Laboratory Tests Common lab equipments and apparatus like test tubes, auto pipettes, centrifuge, balances, oven, water bath go management of needle stick injury & latest guidelines of
Biochemical Laboratory Tests
Antigens, concept involved in vaccine development and
Innate and adaptive immune systems, immunological m development, role of helper T cells (CD4+ T cells) and cy cells in immune responses, Brief concept of MHC Disorders – Immunodeficiency, autoimmunity & hypers
Cells of the Immune System, types of immune systems humoral components of innate and adaptive immune system the formation of antibodies, types, structure and mecha (Immunoglobulins), primary and secondary response
Biochemical tumor markers, biochemical basis of chem therapy, targeted drug therapy and immunotherapy.





11.2 Describe the preparation of buffers and estimation of pH.	Preparation of buffer –acidic and alkaline. Measuremen
11.3 Describe the chemical components of normal urine.	Chemical constituents of normal urine
11.4 & 11.20 11.4: Perform urine analysis to estimate and determine normal and abnormal constituents.	Physical characteristics and organic constituents of urin Collection of random & 24 hour urine sample Urine Report: Physical characteristics and abnormal cor
11.20: Identify abnormal constituents in urine; interpret the findings and correlate these with pathological states.	Interpretation of Urine Abnormalities
11.5 Describe screening of urine for inborn errors & describe the use of paper chromatography. Club Paper chromatography of amino acid & TLC from competency no 11.16	Urine: Screening of inborn errors. Paper chromatograph
11.6 Describe the principles of colorimetry. (Club spectrophotometry from competency no 11.18)	Colorimeter- Principle, Beer and Lambert's law & applic of spectrophotometry.
11.7,11.8, 11.21 & 11.22 11.7- Demonstrate the estimation of serum creatinine and creatinine clearance	Estimation of serum creatinine, urine creatinine and calc their clinical interpretation.
 11.8- Demonstrate estimation of serum proteins, albumin and A:G ratio 11.21- Demonstrate estimation of glucose, creatinine, urea and total protein in serum. 	Estimation of serum proteins, albumin and calculation of interpretation.
11.22- Calculate albumin: globulin A:G ratio and creatinine clearance	Estimation of plasma glucose, serum urea and their clin
11.9	Estimation of serum total cholesterol and HDL choleste interpretation.





Competency No.	Topics & Subtopics
Demonstrate the estimation of serum total cholesterol and HDL	•
tholesterol 11.10 Demonstrate the estimation of triglycerides	Estimation of serum triglycerides and their clinical inter
11.11 Demonstrate estimation of calcium and phosphorous	Estimation of serum calcium and phosphorus their clini
11.12 Demonstrate the estimation of serum bilirubin	Estimation of serum bilirubin: Total, direct and indirect,
11.13 & 2.2 11.13- Demonstrate the estimation of SGOT/ SGPT	Estimation of SGOT (AST)/ SGPT(ALT) and their clinical i
11.14 Demonstrate the estimation of alkaline phosphatase	Estimation of serum ALP and their clinical interpretation
11.15 Describe & discuss the composition of CSF	Physical characteristics and chemical composition of CS



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11.16 & 11.19 11.16- Observe use of commonly used equipment's/techniques in biochemistry laboratory including: •pH meter •Paper chromatography of amino acid •Protein electrophoresis •TLC, PAGE •Electrolyte analysis by ISE •ABG analyzer •ELISA •Immunodiffusion •Autoanalyser •Quality control	Principle, application and working of following lab equip paper chromatography of amino acids, protein electrop analysis by ISE, ABG analyzer, ELISA, immunodiffusion, a DNA isolation from blood/tissue (Paper chromatography of amino acid ,TLC clubbed with
Competency No.	Topics & Subtopics
DNA isolation from blood/ tissue 11.19 Outline the basic principles involved in the functioning of	
instruments commonly used in a biochemistry laboratory and	

their applications.



11.17	Basis and rational of biochemical tests required in the
Explain the basis and rationale of biochemical tests done in the	- Diabetes mellitus-blood & urine glucose, micro
following conditions:	glycated hemoglobin - (Club with 3.8 & 3.10)
- diabetes mellitus,	
- dyslipidemia,	Dyslipidemia-lipid profile (Club with 4.5 & 4.7)
- myocardial infarction,	Dyshpidenna-lipid prome (club mici 4.5 & 4.7)
- renal failure, gout,	Myocardial infarction –CK, LDH, Troponin (Club
- proteinuria,	- Myocardiai infarction –ck, LDN, Hopoliii (Club
- nephrotic syndrome,	
- edema,	Renal failure & nephrotic syndrome, – BUN, Cre
- jaundice,	cholesterol
liver diseases, pancreatitis, disorders of acid- base	(Club with 3.8 & 3.10)
balance, thyroid disorders.	
balance, thyroid disorders.	- Gout- serum uric acid, synovial fluid analysis
200	(Club with 6.3 & 6.4)
11/2	- liver diseases & Jaundice- LFTs (Club with 6.1) P.
	lipase (Club with 2.5& 7 2.7)
· 19.	The section of the se
"MN File the	Disorder of acid base balance- ABG analysis for pH, pO;
20	base excess (BE) (Club with 6.7,6.8)
	Dase excess (DE) (Gub With G.7,0.0)
	Thyroid disorder – serum free and total T3 & T4
11.18	,
	Spectrophotometer –principle & use
Discuss the principles of spectrophotometry.	
(Clubbed with 11.6)	
Competency No.	
	Topics & Subtopics





Outline the basic principles involved in the functioning of instruments commonly used in a Biochemistry laboratory and their applications. (Clubbed with & 11.6 & 11.16) 11.20 Identify abnormal constituents in urine, interpret the findings and correlate these with pathological states. (Clubbed with 11.4) 11.21 Demonstrate estimation of glucose, creatinine, urea and total protein in serum. (Clubbed with 11.7, 11.8) 11.22 Calculate albumin: globulin (A/G)ratio and creatinine clearance (Clubbed with 11.7, 11.8) 11.23 Energy contents of lipids, carbohydrates & proteins in Calculate energy content of different food Items, identify food items with high and low glycemic index and explain the importance of these in the diet.		
Identify abnormal constituents in urine, interpret the findings and correlate these with pathological states. (Clubbed with 11.4) 11.21 Demonstrate estimation of glucose, creatinine, urea and total protein in serum. (Clubbed with 11.7, 11.8) 11.22 Calculate albumin: globulin (A/G) ratio and creatinine clearance (Clubbed with 11.7, 11.8) 11.23 Calculate energy content of different food Items, identify food items with high and low glycemic index and explain the importance of these in the diet. 11.24 Enumerate advantages and/or disadvantages of use of Advantages of unsaturated fats, disadvantages of satu	Outline the basic principles involved in the functioning of instruments commonly used in a Biochemistry laboratory and their applications.	Instruments commonly used in Biochemistry laboratory
Demonstrate estimation of glucose, creatinine, urea and total protein in serum. (Clubbed with 11.7, 11.8) 11.22 Calculate albumin: globulin (A/G) ratio and creatinine clearance (Clubbed with 11.7, 11.8) 11.23 Calculate energy content of different food Items, identify food items with high and low glycemic index and explain the importance of these in the diet. 11.24 Enumerate advantages and/or disadvantages of use of Advantages of unsaturated fats, disadvantages of saturated fats, dis	Identify abnormal constituents in urine, interpret the findings and correlate these with pathological states. (Clubbed with	•
Calculate albumin: globulin (A/G)ratio and creatinine clearance (Clubbed with 11.7, 11.8) 11.23 Calculate energy content of different food Items, identify food items with high and low glycemic index and explain the importance of these in the diet. 11.24 Enumerate advantages and/or disadvantages of use of Advantages of unsaturated fats, disadvantages of saturated fats, disadvantages of sat	Demonstrate estimation of glucose, creatinine, urea and total protein in serum.	
Calculate energy content of different food Items, identify food items with high and low glycemic index and explain the importance of these in the diet. Advantages of unsaturated fats, disadvantages of saturated fats, disadvantages	Calculate albumin: globulin (A/G)ratio and creatinine clearance	
Enumerate advantages and/or disadvantages of use of	Calculate energy content of different food Items, identify food items with high and low glycemic index and explain the	Energy contents of lipids, carbohydrates & proteins in o
	Enumerate advantages and/or disadvantages of use of	Advantages of unsaturated fats, disadvantages of satur



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Paper wise distribution of topics Year: First MBBS Subject: Biochemistry

Paper	Section	Topics	Competency nos. BI
ı	Α	MCQs on all topics of the paper I	
	B & C	Basic Biochemistry	1.1
		Enzymes	2.1-2.7
		Chemistry & metabolism of carbohydrates	3.1-3.10
		Chemistry & metabolism of lipids	4.1-4.7
		Biological oxidation	6.6
		Xenobiotics	7.5
		Antioxidants &defence system	7.6-7.7
		Nutrition	8.1-8.5
		Extracellular matrix	9.1-9.3
		Oncology , oncogenesis & immunity	10.1-10.5
		Biomedical waste	11.1
		Physical characteristics and chemical	11.15
		composition of CSF	
		Energy contents of lipids, carbohydrates	11.23 & 11.24
		& proteins in common food items,	
		Advantages of unsaturated fats.	
		Disadvantages of saturated and trans	
		fats in food	
		AETCOM- 1.4	
	For long an	swer question and scenario based / application	questions, topics
	will not be	repeated.	
II	Α	MCQs on all topics of the paper II	
	B & C	Chemistry & metabolism of proteins	5.1-5.5
		Integration & starvation	6.1
		Nucleic acid metabolism	6.2-6.4
		Vitamins	6.5
		Water electrolyte balance & acid base	6.7-6.8
		balance	
		Mineral metabolism	6.9-6.10
		Haemoglobin chemistry and metabolism	6.11-6.12



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	Organ function test	6.13-6.15
	Molecular biology	7.1-7.3
	Genetic engineering	7.4
	Urine: Screening of inborn errors.	11.5
	Principle, application and working of following lab equipments/techniques: pH meter, paper chromatography of amino acids, protein electrophoresis, TLC, PAGE, Electrolyte analysis by ISE, ABG analyzer, ELISA, immunodiffusion, auto analyzer, quality control, DNA isolation from blood/tissue	11.16
For long an	swer question and scenario based / application	questions,

For long answer question and scenario based / application questions, topics will not be repeated.





Internal Assessment Biochemisry

Applicable w.e.f August 2019 onwards examination for batches admitted from Ju

S	1	-Exam (Decem	ber)		II-Exam (Marc	h)
	Theory	Practical (Including 05 marks For Journals And Log Book)	Total Marks	Theory	Practical (Including 05 marks For Journals And Log Book)	
1	100	50	150	100	50	
	1	Theory	Practical (Including 05 marks Theory For Journals And Log Book)	(Including 05 marks Total For Journals And Log Book)	Practical (Including 05 marks Theory For Journals And Log Book) Total Marks Theory	Practical (Including 05 marks Theory For Journals And Log Book) Practical (Including 05 marks Total Marks Theory For Journals And Log Book) Practical (Including 05 marks Theory And Log Book) Log Book)

_							
		Preliminary Examinations III-Exam (July)				Remedia	al internal assessme Non - eligible st
							October
	Practical Including 10 Marks for Journal & Log Book				Theory	Practical Inclu 10 Marks Journal & Book	
		200	100	300		200	100



- There will be 3 internal assessment examinations in the academic year. The structure of Preliminar should be similar to the structure of University examination.
- There will be only one additional examination for absent students (due to genuine reason) after ap Committee Constituted for the same. It should be taken after preliminary examination and before su internal assessment marks to the University.
- First internal assessment examination will be held in December, second internal assessment examin in March and third internal assessment examination will be held in July.
- Internal assessment marks for theory and practical will be converted to out of 40. Internal assessment
 Conversion, should be submitted to university by 7th of August.
- 5. The student must secure at least 50% marks for total marks (combined in theory and practical / clin 40% marks in theory and practical separately) assigned for internal assessment in a particular subject i eligible for appearing at the final university examination of that subject. Internal assessment marks wi separate head of passing at the summative examination.
- 6. Remedial internal assessment examination for Non eligible students: Student who were not eligible than 50% combined or less than 40% in any theory or practical, will re appear as repeater student for 8 which will be conducted before Supplementary Exam. His/her internal assessment will be calculated of this Examination marks only. Students who will not be eligible in this Examination will appear with regrepeater student.
 - The internal assessment marks of the remedial examination alone shall be considered and conve



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8. Conversion Formula for calculation of marks in internal assessment examinations

	First IA	Second IA	Third IA (Prelim)	Total	Internal assessment marks: Conversion formula (out of 40)	Eligibility to appear (after conversion or (40% Separately in Combined)	ut o
Theory	100	100	200	400	Total marks obtained 10	16 (minimum)	То
Practical	50	50	100	200	Total marks obtained 5	16 (minimum)	M

9. Conversion formula for calculation of marks in Remedial internal assessment examination

	Remedial Exam (Prelim)	Int. Assess. marks conversion formula (out of 40)	Eligibility to ap Exam. (after of (40% Separately 50% Combine	in The
Theory	200	<u>Total marks obtained</u> 5	16 (minimum)	
Practical	100	Total marks obtained 2.5	16 (minimum)	

While preparing Final Marks of Internal Assessment, the rounding-off marks shall done as illustrated in f

Internal Assessment	Marks Final rounded marks
15.01 to 15.49	15
15.50 to 15.99	16





First Year MBBS Practical Mark's Structure Internal Assessment Examinations I & II (Application M.B.B.S Course from Academic Year 2019-20 & onwards)

	Biochemistry										
	Practical										
Seat No.	Quantitative Experiment	Quantitative Experiment/Urine organic/Urine Report/Quality Control/Interpolation of lab Report /Interpolation of Special Technique	Spots	Journal/ Logbook	Oral/V						
	А	ми. в	с	D	E						
Max. Marks	15	15	5	5	10						



First Year MBBS Practical Marks Structure (Prelin

(Applicable w.e.f August 2019 onwards examination for batches admitted from June 201

Biochemistry

			4 954					
Seat No	Case Based Quantitative Estimation	Urine Report/ Quantitative estimation	Quality Control	Interpretation of lab Reports & special techniques (Minimum 2 Interpretation)	Spots	Journal & Logbook	Practical Total	
	Α	В	C	D	E	F	G	
Max. Marks	25	15	200	20	10	10	90	
		Lite	5					

(Please Note - The above examination pattern will be applicable to the students admitted from Academic Year 2019-20 and on Colleges vide University letter No MUHS /X-1 /UG /1692 /2020 Date: 28/02/2020)



First Year MBBS Practical Marks Structure (MUHS E

(Applicable w.e.f August 2019 onwards examination for batches admitted from June 201

Biochemistry

Seat No	Case Based Quantitative Estimation	Urine Report/ Quantitative estimation	Quality Control	Interpretation of lab Reports & special techniques (Minimum 2 Interpretation)	Spots	Practical Total	Viva Voce/ Oral
	Α	В	8	D	E	F	G
Max. Marks	25	15	10	20	10	80	20
		N.Y.					
		M.					

(Please Note - The above examination pattern will be applicable to the students admitted from Academic Year 2019-20 and onw Colleges vide University letter No MUHS /X-1 /UG /1692 /2020 Date: 28/02/2020)

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MAHARASHTRA UNIVERSITY OF HEALTH SCIENCES, NASHIK FORMAT / SKELETON OF QUESTION PAPER

1.	C	ourse a	ind Y	ear		First													2.	Sui	bject Code		: /	Appendix - a
,	-	- Marie	men	-0.					f. Sept. 2					inatte	ions)									
3.	34	ubject			- 1	Anato	my/	Ph	iysiolog	y/Bi	ocher	mistr	ry											
l,			(II)						Total	-	- 1	-00		4 1	Total Time			TE.	-	7.	- (mg)		- I	D- 300/
4.	P	aper :			1	1		5.	Total Ma	arks	: 1	.00		6.	Total Time	÷	3	Hrs.	_		emu. (PS)			Rs. 300/-
										-											emu. (PM)		: 1	Rs. 350/-
9.	W	Veb Pat	tem		: []		10.	Web Ske	eleton	: []		11.	Web Syllabus	;	[]	12	2. We	eb Old QP		: []
Γ.		-									SE	ст	ION	"A"	мсо									
I	u.str	ruction	13.7		1)	Fil	l 🗨	(da	irk) the o	ирргор					elow the que	stion i	mun	mber once	only	7				
					2)				lack ball tion carri															
					4)								mari	ks if	he/she over	writes	S, 5	strikes ou	t or	puts	white ink	on	the	circle once filled
					_	(da	arken	ted)																
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(applicable w.e.f. Sept. 2020& onwards examinations)

Course and : First MBBS

Year

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· Appendix - a

2. Subject Code

MAHARASHTRA UNIVERSITY OF HEALTH SCIENCES, NASHIK FORMAT / SKELETON OF QUESTION PAPER

3. Subje	ect (PSP)	: /	naton	ny / P	hysiolo	ogy / l	Bioch	emist	ry							
	(TT)	:														
. Paper	r:	: 1	I	5.	Total	Marks	:	100	6.	Total Time	:	3 Hrs.	_	Remu. (PS)		Rs. 300/-
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2	Brief an	swer (anestio	ns (Ar	v Ten o	ut of F										(10x 2= 20)
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3.	Short A															(8x5=40)
	Minimi	um 2 8	sAQs s	hould	be Case	Based	Ques	tions/	Clinical	ly applied Que	stions					
4.		a)	b)	c) d)	e)	f)	g) l	h) i)							(2x 10=20)
	Long /	Answe	r Ques	tions (Any Tw	o out	of The	ee)								
	a)	b)	c)		-											
		,	- ,													

Note: All questions should be structured. Wherever necessary, split up of marks should be specified.



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MAHARASHTRA UNIVERSITY OF HEALTH SCIENCES, NASHIK MARKLIST FOR PRACTICAL / ORAL / VIVA VOCE

(Summer / Winter - 20...Exam (MBBS UG Courses)

(Applicable for batch admitted in M.B.B.S Course from Academic Year 2019-20 & onwards)

Course : FIRST MBBS Subject : Biochemistry

CENTRE: Marks: (Practical = Practical/Clinical + Viva) Min. 50 Max. 100

Date: / /20 Batch:

		Oral/Viva	Total					
Seat No.	Case Based Quantitative Estimation	Urine Report/ Quantitative estimation	Quality Control	Interpretation of lab Report & special techniques (Minimum 2 interpretations)	Spots	Practical (Total)	Viva Voce/Oral Total	Practical/Viva Total Marks
	A	В	с	D	E	F	G	н
Max. Marks	25	15	10	20	10	80	20	100
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Note: Both Examiners should jointly conduct practical examination for each student.

Verified above entries from Answerbooks and we hereby certify that the marks entered against each Seat Number are found correct.

	NAME OF EXAMINER	COLLEGE	SIGNATURE WITH DATE				
1			Convenor				
2			Internal				
3			External				
4			External				





BOOKS RECOMMENDED: TEXT BOOKS;

- Medical Biochemistry U.Satyanarayan.
- Biochemistry for Medical students by D.M.Vasudevan & Shree Kumari.
- Medical Biochemistry by M.N. Chatterjea and Rana Shinde.
- Text Book of Medical Biochemistry by Ramakrishnan, Prasannan & Rajan.
- Medical Biochemistry by Debajyoti Das.
- Biochemistry by A.C.Deb.

REFERENCE BOOKS:

- Biochemistry by Pankaja Naik
- 2. Harper's Biochemistry.
- 3. Medical Biochemistry by N.V.Bhagwan.
- Biochemistry by L.Stryer.
- Biochemistry by Orten & Neuhans.
- 6.Text Book of Biochemistry for Medical Student-8th Edition-16 By- DM Vasudevan

