(LK 4251)

FEBRUARY 2017

Sub Code: 4251

B.PHARM. EXAMINATION FIRST YEAR **PAPER I – PHARMACEUTICAL INORGANIC CHEMISTRY**

Q.P. Code: 564251

Maximum: 100 Marks

I. Elaborate on:

Time: Three hours

- 1. a) What are the sources of impurities in pharmaceutical substances? Explain the principle for the limit test for sulphates.
 - b) What are antimicrobials? Write the preparation and assay of chlorinated lime and boric acid.
- 2. a) What are gastrointestinal agents? Classify them with examples. Describe about qualities of an ideal antacid and combination therapy of antacids.
 - b) Write about theory of coordination compounds with special reference to application in pharmacy and pharmaceutical analysis.

II. Write notes on:

- 1. What are medicinal gases? Write properties and assay of nitrous oxide.
- 2. What are radiopharmaceuticals? Write their clinical applications.
- 3. What are dentrifices? Write the preparation and assay of calcium carbonate.
- 4. Write about physiological acid-base balance and its importance.
- 5. Discuss about the development of periodic table on the modern concept of atomic structure and its importance.
- 6. What are saline cathartics? Write the preparation and assay of magnesium sulphate.
- 7. Write the method of preparation, assay and uses of ferrous sulphate.
- 8. Write the principle involved in the limit test for lead.

III. Short answers on:

- 1. Write about protophilic solvents with examples.
- 2. Give the physiological role of potassium.
- 3. Write a note on complexometry titration with examples.
- 4. Define acidifiers with examples.
- 5. Write about radio opaque contrast medium.
- 6. Write the structure and uses of dimercaprol.
- 7. Define antidotes with examples.
- 8. Write a note on alum.
- 9. Define sclerosing agents with examples.
- 10. Write about calamine.

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 $(8 \times 5 = 40)$

(10 x 2 = 20)

 $(2 \times 20 = 40)$