

Code: R7312301

R7

B.Tech III Year I Semester (R07) Supplementary Examinations, May 2012

TRANSPORT PHENOMENA IN BIO PROCESSES

(Biotechnology)

Time: 3 hours

Max Marks: 80

Answer any FIVE questions

All questions carry equal marks

- 1 (a) Give a brief account on Non-Newtonian fluids.
(b) Describe the different factors effecting broth viscosity.
- 2 (a) A fermentation broth with viscosity 10^{-2} Pa s and density 1000 Kg m^{-3} is agitated in 50 m^3 baffled tank using a marine propeller 1.3 m is diameter. Calculate the power required for a stirrer speed of 4 S^{-1} . The tank is fitted with four baffles. (Power number versus Reynolds number graph should be provided during the examination)
(b) Explain about interface transport in isothermal system.
- 3 (a) The temperature at the inside surface of a hollow silver sphere is 85° C and the outside surface, 15° C . The inside diameter is 5 cm and the outside diameter is 7.5 cm. The value of K of silver is 425 W/mK . Find the rate of heat flow.
(b) In what way are Newton's law of viscosity and Fourier's law of heat conduction similar? Dis similar.
- 4 A $400 \times 400 \text{ mm}$ copper slab 5 mm thick at a uniform temperature of 250° C suddenly has its surface temperature lowered to 30° C . Find the time at which the slab temperature becomes 90° C . $\rho = 9000 \text{ Kg/m}^3$, $c = 0.38 \text{ KJ/Kg}^{\circ} \text{ K}$, $k = 370 \text{ W/m}^{\circ} \text{ K}$, $h = 90 \text{ W/m}^2 \text{ K}$. All terms with usual notations.
- 5 (a) Explain in detail about two film theory with neat sketch.
(b) Determine the diffusivity of CO through a mixture of N_2 and O_2 in which the concentration of CO is essentially zero. The gas mixture will be at 25° C and 2 atm pressure. Other data: $D_{\text{CO}-\text{O}_2} = 0.185 \frac{\text{cm}^2}{\text{s}}$ at 273° K , 1 atm: $D_{\text{CO}-\text{N}_2} = 0.192 \frac{\text{cm}^2}{\text{s}}$ at 288° K .
- 6 Solvent A is evaporating out of a coat of lacquer on a plane surface exposed to a tangential stream of non-condensable gas B. At a given point on the surface the gas phase mass transfer coefficient at the prevailing average fluid properties is given as $0.1 \text{ lb-mole/hr ft}^2$; the Schmidt number is = 2.0. The interfacial gas composition is $X_{\text{AO}} = 0.8$. Estimate the local rate of evaporation using the stagnant film model.
- 7 Calculate the maximum rate of absorption of O_2 in a fermenter from air bubbles at 1 atm abs pressure having diameters of $100 \mu\text{m}$ at 37° C into water having a zero concentration of dissolved O_2 . The solubility of O_2 from air in water at 37° C is $2.26 \times 10^{-4} \text{ Kgmol O}_2/\text{m}^3$. The diffusivity of O_2 in water at 37° C is $3.25 \times 10^{-9} \text{ m}^2/\text{s}$. Agitation is used to produce the air bubbles.
- 8 (a) Explain briefly about how oxygen mass transfer and $k_L a$ can limit the biomass density in ferementors.
(b) Explain briefly about the dynamic method for the estimation of mass transfer coefficient.
