

Code: R7100305



B.Tech I Year (R07) Supplementary Examinations December/January 2015/2016

ENGINEERING CHEMISTRY

(Mechanical Engineering) (For 2008 regular admitted batch only)

Time: 3 hours

Max. Marks: 80

Answer any FIVE questions All questions carry equal marks

- 1 (a) What is meant by sterilization of water? Explain how sterilization of water is carried out by using chlorine and ozone.
 - (b) How is hardness of water expressed? What are the various units employed? Explain the interconversion.
- 2 (a) Why is hardware harmful to boiler? Describe the causes and harmful effects of scale formation.
 - (b) Calculate the amount of soda and lime required to soften 15,000 liters of water which showed the following analysis: NaCl = 15.5 PPM, KCl = 25 PPM, Mg(HCO₃)₂ = 3.30 PPM, CO₂ = 5.5 PPM, MgSO₄ = 120 PPM, CaCl₂ = 22 PPM, Ca(HCO₃)₂ = 15 PPM, suspended matter = 15 PPM, organic impurities = 3.5 PPM.
- 3 (a) Explain how heterogeneity of metal increases the rate of corrosion.
 - (b) What are corrosion inhibitors? Discuss anodic and cathodic inhibitors with examples.
- 4 (a) Explain any three types of surface preparations.
 - (b) Write short notes on phosphate and anodized coatings.
- 5 (a) Distinguish between addition and condensation polymerization.
 - (b) Write the properties and uses of different types of silicones.
- 6 (a) What are refractories? How important are the properties, refractories under load and thermal conductivity for industrial applications?
 - (b) Explain the characteristics and applications for thermal and electrical insulators.
- 7 (a) Describe the various types of lubrication.
 - (b) Write short notes on:
 - (i) Cloud point and pour point.
 - (ii) Fire and flash point.
- 8 (a) Compare petrol and diesel oil with respect to:
 - (i) Knocking characteristics.
 - (ii) Octane number and Cetane number.
 - (b) Describe the manufacture of gasoline by Bergius process.

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