R09

Set No. 2

II B.Tech I Semester Examinations, November 2010 METALLURGICAL ANALYSIS Metallurgy And Material Technology

Time: 3 hours Max Marks: 75

> Answer any FIVE Questions All Questions carry equal marks

- 1. Explain and Give the classification of various methods used in Metallurgical Analysis. |15|
- 2. (a) What are the different electrodes and samples used in spectroscopy and how they are prepared?
 - (b) Draw neatly the schematic diagram of a photo electric spectrometer. [15]
- 3. Explain the steps involved in conventional solution methods for qualitative analysis of Bauxite ore. |15|
- 4. Explain about the Hilger spekker Absorption meter with a neat sketch. [15]
- 5. What properties are improved by the addition of molybdenum and vanadium to alloy steel and how they are determined by gravimetric method or ignition method. [15]
- 6. Differentiate between

Code No: A109211803

- (a) Ionic mechanism and Electrode processes.
- (b) Electrolysis process and Free radical mechanism.
- (c) Conductometry and polarography.

[15]

- 7. (a) What are the important non-ferrous alloys which are used in commercial applications.
 - (b) Explain about the analysis of lead and beryllium in bronzes. [15]
- 8. Explain about the following:
 - (a) Determination of reducing power of ores.
 - (b) Determination of oxidising power of ores.
 - (c) Assaying of ores.

[15]

Code No: A109211803

R09

Set No. 4

II B.Tech I Semester Examinations, November 2010 METALLURGICAL ANALYSIS Metallurgy And Material Technology

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. Discuss about the Electromigration in solid metals and Alloys. [15]
- 2. Explain the estimation of manganese dioxide by titration method from manganese ore. [15]
- 3. (a) Define precipitation and list out the conditions of precipitation.
 - (b) What are the rules of precipitation and give with example. [15]
- 4. Discuss the physico chemical principles involved and equipment required in potentiometry and list out the applications of potentiometry. [15]
- 5. Explain in detail the estimation of molybdenum in alloy steels by Jones reductor method. [15]
- 6. Differentiate between the following:
 - (a) Persulphate Arsenite method and chlorate or Ford and Williams method.
 - (b) Strohlein method and Eggertz method. [15]
- 7. What are the conventional solution methods? Why it is used only for analysis of ores, slags, refractories and Give their advantages and disadvantages. [15]
- 8. Discuss the physico chemical principles involved and equipment required in Duplication method of colorimetry and what are its applications. [15]

Code No: A109211803

R09

Set No. 1

II B.Tech I Semester Examinations, November 2010 METALLURGICAL ANALYSIS Metallurgy And Material Technology

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. What are the major constituents of limestone and discuss the analysis of lime, Magnesia and silica from limestone. [15]
- 2. What are the advantages and disadvantages of gravimetric method which is used for Molybdenum estimation from alloy steel and explain. [15]
- 3. Explain the fundamental physico chemical principles involved in Instrumental Analysis. [15]
- 4. (a) What is a polarograph and polarogram, explain the residual current, migration current and diffusion current.
 - (b) Describe the important Instrumental features of a modern atomic absorption spectrophotometer. [15]
- 5. Explain the methods of quantitative analysis used for estimating ores, elements and alkali metals. [15]
- 6. Explain in detail potassium permanganate method to determine total iron in cast irons. [15]
- 7. What are the important ores of silver and explain the different methods used for estimating of silver from their respective ores. [15]
- 8. Explain in detail about the dissolution of the ores with examples. [15]

Code No: A109211803

R09

Set No. 3

II B.Tech I Semester Examinations, November 2010 METALLURGICAL ANALYSIS Metallurgy And Material Technology

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. Explain the steps involved in conventional solution methods for qualitative analysis of Iron ore. [15]
- 2. What are the different types of furnaces which are used for fire assaying and explain in detail about the furnaces? [15]
- 3. Define Metallurgical Analysis and explain the scope and importance of Metallurgical Analysis. [15]
- 4. (a) List out the important applications of flame photometric methods.
 - (b) Differentiate between standard addition method and calibration curve method.
 [15]
- 5. Explain the determination of Nickel in steels and Its Alloys by spectrophotometry and list out the advantages and disadvantages of this process. [15]
- 6. Write short notes on any two:
 - (a) Cyanide method used for nickel estimation
 - (b) Chromium estimation by electrolysis process
 - (c) Tungsten estimated by gravimetric method. [15]
- 7. Discuss in detail about the analysis of Bismuth and iron from solders by gravimetric and volumetric methods. [15]
- 8. (a) How is Spectroscopy used in the analysis of various elements in an alloy?
 - (b) Outline the principles involved in Potentiometric titrations.
 - (c) List out the different types of indicators and electrodes used in Potentiometric titrations. [15]
