

Code No: R07A1BS07

R07**Set No. 2****I B.Tech Examinations, December 2010****ENGINEERING CHEMISTRY****Common to Mechanical Engineering, Mechatronics, Production
Engineering, Automobile Engineering****Time: 3 hours****Max Marks: 80****Answer any FIVE Questions
All Questions carry equal marks**

1. (a) Define lubricant and lubrication.
(b) Explain the mechanism of lubrication behind delicate machines like watches, and clock etc.
(c) What is meant by oiliness of a lubricant? How can this be improved? [4+6+6]
2. (a) Calculate the amounts of lime (85% pure) and soda (95% pure) required to soften a million litres of water containing the following constituents per litre $\text{Ca}(\text{HCO}_3)_2$ -243 mg/L; $\text{Mg}(\text{HCO}_3)_2$ -73mg/L; CaSO_4 -102 mg/L; MgCl_2 -95mg/L; and NaCl - 500mg/L.
(b) Explain the requisites of water used for paper mills, textile and dyeing industry, sugar industry and beverage industry. [8+8]
3. (a) What is pyrometric cone equivalent? How it is determined for a refractory? What is its significance?
(b) Write a short note on:
 - i. porosity
 - ii. Thermal Conductivity
 - iii. Dimensional Stability.
 - iv. Strength [8+8]
4. (a) Describe the rusting of iron using acid corrosion theory.
(b) Explain electrochemical theory of corrosion with suitable examples. [4+12]
5. (a) Explain the preparation, properties and uses of Bakelite.
(b) Describe with a neat sketch, the process of compression moulding. [10+6]
6. (a) Describe the principle, process of electroplating with an example.
(b) Why utensils coated with zinc are not used for storing food stuff whereas tin coated utensils are used? [12+4]
7. Write a short notes on the following:
 - (a) Break - point chlorination
 - (b) Dissolved oxygen
 - (c) Hardness of water

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(d) Sedimentation and coagulation.

[4+4+4+4]

8. (a) What are the different varieties of coal? Give the physical state, percentage of carbon, calorific value, and application in each case.
- (b) What volume of air is required for the complete combustion of the followings:
- i. 10 m^3 of Hydrogen
 - ii. 5 m^3 of Methane.

[8+8]

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R07**Set No. 4****I B.Tech Examinations, December 2010****ENGINEERING CHEMISTRY****Common to Mechanical Engineering, Mechatronics, Production
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(b) Explain the requisites of water used for paper mills, textile and dyeing industry, sugar industry and beverage industry. [8+8]
2. (a) Explain the preparation, properties and uses of Bakelite.
(b) Describe with a neat sketch, the process of compression moulding. [10+6]
3. Write a short notes on the following:
(a) Break - point chlorination
(b) Dissolved oxygen
(c) Hardness of water
(d) Sedimentation and coagulation. [4+4+4+4]
4. (a) What is pyrometric cone equivalent? How it is determined for a refractory? What is its significance?
(b) Write a short note on:
i. porosity
ii. Thermal Conductivity
iii. Dimensional Stability.
iv. Strength [8+8]
5. (a) Describe the principle, process of electroplating with an example.
(b) Why utensils coated with zinc are not used for storing food stuff whereas tin coated utensils are used? [12+4]
6. (a) What are the different varieties of coal? Give the physical state, percentage of carbon, calorific value, and application in each case.
(b) What volume of air is required for the complete combustion of the followings:
i. 10 m^3 of Hydrogen
ii. 5 m^3 of Methane. [8+8]

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7. (a) Describe the rusting of iron using acid corrosion theory.
(b) Explain electrochemical theory of corrosion with suitable examples. [4+12]
8. (a) Define lubricant and lubrication.
(b) Explain the mechanism of lubrication behind delicate machines like watches, and clock etc.
(c) What is meant by oiliness of a lubricant? How can this be improved?[4+6+6]

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R07**Set No. 1****I B.Tech Examinations, December 2010****ENGINEERING CHEMISTRY****Common to Mechanical Engineering, Mechatronics, Production
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1. (a) Explain the preparation, properties and uses of Bakelite.
(b) Describe with a neat sketch, the process of compression moulding. [10+6]
2. (a) Describe the rusting of iron using acid corrosion theory.
(b) Explain electrochemical theory of corrosion with suitable examples. [4+12]
3. (a) Calculate the amounts of lime (85% pure) and soda (95% pure) required to soften a million litres of water containing the following constituents per litre $\text{Ca}(\text{HCO}_3)_2$ -243 mg/L; $\text{Mg}(\text{HCO}_3)_2$ -73mg/L; CaSO_4 -102 mg/L; MgCl_2 -95mg/L; and NaCl - 500mg/L.
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i. 10 m^3 of Hydrogen
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5. (a) Define lubricant and lubrication.
(b) Explain the mechanism of lubrication behind delicate machines like watches, and clock etc.
(c) What is meant by oiliness of a lubricant? How can this be improved? [4+6+6]
6. Write a short notes on the following:
(a) Break - point chlorination
(b) Dissolved oxygen
(c) Hardness of water
(d) Sedimentation and coagulation. [4+4+4+4]
7. (a) What is pyrometric cone equivalent? How it is determined for a refractory? What is its significance?
(b) Write a short note on:
i. porosity

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- ii. Thermal Conductivity
- iii. Dimensional Stability.
- iv. Strength

[8+8]

8. (a) Describe the principle, process of electroplating with an example.
- (b) Why utensils coated with zinc are not used for storing food stuff whereas tin coated utensils are used? [12+4]

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R07**Set No. 3****I B.Tech Examinations, December 2010****ENGINEERING CHEMISTRY****Common to Mechanical Engineering, Mechatronics, Production
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2. (a) Define lubricant and lubrication.
(b) Explain the mechanism of lubrication behind delicate machines like watches, and clock etc.
(c) What is meant by oiliness of a lubricant? How can this be improved? [4+6+6]
3. (a) Describe the principle, process of electroplating with an example.
(b) Why utensils coated with zinc are not used for storing food stuff whereas tin coated utensils are used? [12+4]
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(b) What volume of air is required for the complete combustion of the followings:
i. 10 m^3 of Hydrogen
ii. 5 m^3 of Methane. [8+8]
7. (a) What is pyrometric cone equivalent? How it is determined for a refractory? What is its significance?
(b) Write a short note on:
i. porosity
ii. Thermal Conductivity
iii. Dimensional Stability.
iv. Strength [8+8]

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8. Write a short notes on the following:

- (a) Break - point chlorination
- (b) Dissolved oxygen
- (c) Hardness of water
- (d) Sedimentation and coagulation.

[4+4+4+4]

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