

I B.Tech II Semester Supplementary Examinations, August 2014 ENGINEERING PHYSICS -II

 (Common to Civil Engineering, Electrical & Electronics Engineering, Mechanical Engineering, Electronics & Communication Engineering, Computer Science & Engineering, Chemical Engineering, Electronics & Instrumentation Engineering, Bio-Medical Engineering, Information Technology, Electronics & Computer Engineering, Aeronautical Engineering, Bio-Technology, Automobile Engineering, Mining and Petroliem Technology)

Time: 3 hours

Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks ****

- 1. (a) What are matter waves. Explain their properties
 - (b) Show that the wave length λ associated with an electron of mass, m and Kinetic energy, E is given by $\lambda = \frac{h}{\sqrt{2mE}}$, where h is planck's constant.
 - (c) Calculate the wavelength associated with an electron with energy 2000eV $[8{+}4{+}3]$
- 2. (a) Explain briefly the classical free electron theory of metals.
 - (b) Derive an expression for electrical conductivity on the basis of classical free electron theory. [8+7]
- 3. (a) State and explain Bloch theorem in a periodic potential field.
 - (b) Explain the classification of solids into conductors, semiconductors and insulators. [7+8]
- 4. (a) What is ferromagnetic Hysteresis? Explain retentivity and coercivity. Explain ferromagnetic hysteresis on the basis of Domain theory?
 - (b) Classify & Distinguish ferromagnetic materials on the basis of the hysteresis loop. [8+7]
 - (a) What is the Superconductivity? Explain Meissner effect.
 - (b) Mention few industrial applications of superconductors.
 - (c) Calculate the critical current which can flow through a long thin superconducting wire of aluminium of diameter 10^{-3} m. The critical magnetic field for aluminium is 7.9×10^{3} amp/m. [6+5+4]
- 6. (a) What is ionic polarizability? Derive an expression for the ionic polarizability. Explain frequency dependence of ionic polarizability.
 - (b) If all the molecular dipoles in a 1.0 cm radius water drop are pointed in the same direction, calculate the intensity of polarization. Dipole moment of the water molecule is $6 \ge 10^{-30}$ C-m. [9+6]



- 7. (a) Derive an expression for carrier concentration in intrinsic semiconductor.
 - (b) The forbidden gap in pure silicon is 1.1eV. Compare the number of conduction electrons at temperatures 37^{0} C and 27^{0} C. [11+4]
- 8. (a) What are nanomaterials? Why do they exhibit different properties?
 - (b) Write the applications of Nano Technology. [9+6]





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- 1. (a) What is quantum computer? Explain the advantages of quantum computing over classical computation.
 - (b) Explain briefly about multiple qu bit. [8+7]
- 2. (a) Mention different mechanisms responsible for electrical resistance in metals.
 - (b) Find the temperature at which there is 1% probability of a state with an energy 0.5 eV above Fermi energy level will be occupied [8+7]
- 3. (a) Discuss the motion of an electron in a periodic potential field and explain the formation of energy bands.
 - (b) What is the effective mass of an electron? Derive the expression for the effective mass of an electron moving in energy bands of a solid. [7+8]
- 4. (a) Write the classification of Magnetic Meterial.
 - (b) Explain hysteresis loop observed in ferromagnetic materials.
 - (c) The area of the hysteresis loop between B and H is 100 m². Each unit space along the vertical axis represents 0.001 Wb/m² and each unit space along the horizontal represents 40A/m. Determine the Hysteresis loss per cycle.

[6+5+4]

- (a) Explain the following:(i)Flux quantization.(ii)Penetration depth.
- (b) Explain critical magnetic field of a superconductor as a function of temperature.
- (c) For lead, Tc = 7.2K. Calculate the ratio of penetration depth at 3K to that at 0K. [6+5+4]
- 6. (a) Define dielectric polarizability and susceptibility.
 - (b) Derive the expression for electronic polarizability.

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(c) The dielectric constant of helium, measured at 0^0 C and at one atmosphere is 1.0000684; under these conditions the gas contains 2.7×10^{25} atoms/m³. Calculate the atomic radius. [4+7+4]

Set No. 2

- 7. (a) Distinguish between conductors, insulators and semiconductors.
 - (b) The intrinsic carrier density at room temperature in Ge is 2.37X 10^{19} /m³. If the electron and hole mobilities are 0.38 and 0.18 m² /v/s respectively, calculate the resistivity. [9+6]
- 8. (a) What is Quantum Confinement? Explain density of states for various types of Quantum Confinement.
 - (b) Explain Ball Milling process and Sol-Gel process in nanomaterials. [8+7]





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- 1. (a) Explain the Physical significance of Wave function.
 - (b) Derive time dependent Schrodinger wave equation. [7+8]
- 2. (a) Explain briefly the classical free electron theory of metals.
 - (b) Derive an expression for electrical conductivity on the basis of classical free electron theory. [8+7]
- 3. (a) State and explain Bloch theorem in a periodic potential field.
 - (b) Explain the classification of solids into conductors, semiconductors and insulators. [7+8]
- 4. (a) Define the following terms (i)Magnetic permeability
 - (ii) Magnetic susceptibility
 - (iii) Coercivity and
 - (iv) retentivity
 - (b) Derive the relation between B, H and I
 - (c) A magnetic material has intensity of magnetization of 1550 A/m and flux density of 0.0022Wb/m². Calculate the magnetizing force and relativity permeability of the material. [8+4+3]
 - (a) Write general properties of superconductors.
 - (b) Draw the magnetization curves for Type-I & Type-II superconductors and mention different regions.
 - (c) The lead material works as superconductor at a temperature of $T_c = 7.26$ K. If $H_0 = 8 \ge 10^5$ A/m find critical magnetic field at 5K. [6+5+4]
- 6. (a)Deduce the relation between polarizability (α) and relative permittivity \in_r . (b) Explain briefly the Ferro electricity and Piezoelectricity. [7 + 8]

- 7. (a) Explain Hall Effect. Show that for a p-type semiconductor the hall coefficient $R_H = 1/ne$.
 - (b) write any four applications of Hall effect.
- 8. (a) What are the types of Carbon nanotubes? Mention their properties(b) How the physical and chemical properties of nano-particles vary with their size?

[10+5]

[11+4]

Set No. 3

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- 1. (a) What is a quantum computer? Explain briefly how it works.(b) What is Bloch's Sphere and describe any two properties of it. [8+7]
- 2. (a) Explain briefly the classical free electron theory of metals.
 - (b) Derive an expression for electrical conductivity on the basis of classical free electron theory. [8+7]
- 3. (a) Explain the formation of allowed and forbidden energy bands on the basis of the Kronig- Penny model.
 - (b) Write short notes on "effective mass" (\mathbf{m}^*) of an electron. [8+7]
- 4. (a) Define the following (i)Magnetic moment
 - (ii) Intensity of magnetization
 - (iii) Magnetizing force
 - (iv) magnetic flux density
 - (b) Explain the origin of magnetic moment at the atomic level.
 - (c) Find the relative permeability of a ferromagnetic material if a field of strength 110 A/m produces a magnetization of 3300 A/m. [8+5+2]
 - (a) Distinguish the variation of resistivity with temperature in normal and superconductors.
 - (b) Mention important property changes that occur in materials when they change from normal to superconducting state.
 - (c) Distinguish energy gap in semiconductors and superconductors. [4+7+4]
- 6. (a) Describe the various of processes of dielectric polarization.
 - (b) Derive expressions for electronic & ionic polarizabilities. [9+6]
- 7. (a) Derive an expression for carrier concentration in intrinsic semiconductor.
 - (b) The forbidden gap in pure silicon is 1.1eV. Compare the number of conduction electrons at temperatures 37^{0} C and 27^{0} C. [11+4]

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[7+8]

- 8. (a) Write a detailed note on nano science and nanotechnology
 - (b) Describe briefly the fabrication methods of nanomaterials

