CT Inst. of E

Roll No.

Total No. of Questions: 09

B.Tech. (ME) (Sem.-3rd)

**ENGINEERING MATERIALS & METALLURGY** 

Subject Code: BTME-306 (2011 Batch)

Paper ID: [A1143]

Paper ID : [A114;

Time: 3 Hrs.

Max. Marks: 60

Total No. of Pages: 02

#### INSTRUCTION TO CANDIDATES :

- SECTION-A is COMPULSORY consisting of TEN questions carrying TWO marks each.
- SECTION-B contains FIVE questions carrying FIVE marks each and students has to attempt any FOUR questions.
- SECTION-C contains THREE questions carrying TEN marks each and students has to attempt any TWO questions.

## SECTION-A

### I. Write briefly:

- a) What is FCC Structure?
- b) Show that Atomic Packing Factor for BCC is 0.68.
- c) What is meant by atomic bonding? Name the various atomic bonds in solids.
- d) What is a Gibbs Phase Rule?
- e) What is retained austenite?
- f) What is meant by limited solubility in solid state?
- g) Define Harden ability.
- h) Why annealing is done?
- i) Write two applications of steel containing Mo.
- j) Which properties of tool steel will be affected by chromium?

# SECTION-B

- Explain the method of crystallographic notat the atomic planes of close packed hexagona
- 3. Explain the theories associated with the plas
- What are various types of equilibrium diagrams.
  Explain any one with detail.
- What is meant by surface hardening? Explain detail.
- 6. What should be the properties of a cutting t elements suitable for the same?

#### SECTION-C

- Explain the principle and applications of tir curves.
- 8. What is Diffusion Mechanisms? Explain the
- 9. How the Normalizing is done? Explain the treatment and how these can be removed?

