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Enrolment No. \_\_\_\_\_ Seat No.: \_\_\_\_\_

## CHIADAT TECHNOLOGICAL LINIVED SITV

B.Ph SEMESTER- II• EXAMINATION – SUMMER - 2018				
Sub	iect (	Code: 220003 Date:24/05/2018	Date:24/05/2018	
•	•	Name: Pharmaceutical Chemistry-II		
Time: 10:30 AM TO 01:30 PM  Instructions:  Total Marks: 80				
		tempt any five questions.		
		ake suitable assumptions wherever necessary.		
3.	. Fig	gures to the right indicate full marks.		
Q.1	(a)	What is adsorption? Write note on Gibbs adsorption isotherm.	06	
	<b>(b)</b>	Write a note on Debye Huckel theory.	05	
	(c)	Differentiate between Thermochemistry and Photochemistry.	05	
Q.2	(a)	State and explain first law of thermodynamics with various modifications.	06	
	<b>(b)</b>	Discuss the methods of determination of order of reactions.	05	
	(c)	Define and explain surface tension, optical activity and refractive index.	05	
Q.3	(a)	State the distribution law. Explain partition coefficient with suitable example in detail.	06	
	<b>(b)</b>	Write a note on joule Thomson effect.	05	
	(c)	Write in detail about the factors affecting viscosity.	05	
Q.4	(a)	Define Colligative properties. Enlist different types of it and explain in detail any two.	06	
	<b>(b)</b>	Define $C_P$ and $C_V$ . State their relation.	05	
	(c)	Write a short note on Carnot cycle.	05	
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Q.5	(a)	What is radioactivity? What are the different types of radiations emitted by radioactive substances? Explain it in detail.	06	
	<b>(b)</b>	Write an informative note on Geiger Muller counter.	05	
	(c)	State the applications of radioactivity.	05	
Q. 6	(a)	Define half-life of a reaction. Prove that half-life of a first order reaction is independent of the initial concentration.	06	
	<b>(b)</b>	State and explain Henry's law.	05	
	(c)	Explain important characteristics of Enzyme catalysis.	05	
Q.7	(a)	Define viscosity. Explain in detail any two methods for measurement of viscosity.	06	
	<b>(b)</b>	Define surface tension. Explain drop weight method for the determination of surface tension.	05	
	(c)	What is phase rule? Discuss water system with reference to phase rule.	05	