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BE - SEMESTER-III (NEW) - EXAMINATION - SUMMER 2018

Subject Code:2130506 Date:18/05/2018

Subject Name: Chemical Engineering Materials

Time:10:30 AM to 01:00 PM Total Marks: 70

Instructions:

1.	Attempt	all d	questions.
	LICULIA	an d	questions.

- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

Q.1	(a)	Explain Atomic structure with diagram.	03
	(b)	List out different X-ray diffraction methods and explain any one.	04
	(c)	Explain Bohr's Model with its drawbacks.	07
Q.2	(a)	Write a note on conductors.	03
	(b)	Explain primary bonding in solids.	04
	(c)	Define Refractories & explain its classification.	07
		OR	
	(c)	Define Abrasives & explain its classification.	07
Q.3	(a)	Discuss basic process of powder metallurgy.	03
	(b)	State different types of Oils and their properties	04
	(c)	Explain coordination number with suitable example. Explain along with	07
		crystal defects.	
0.2	(-)	OR	02
Q.3	(a)	Define: Ionic conductivity, semiconductors and insulators	03
	(b)	Explain electrical and magnetic property of engineering materials.	04
	(c)	Explain applications, merits and demerits of powder metallurgy.	07
Q.4	(a)	Define: Refractories, Abrasives and Cement	03
	(b)	Give classifications of paints and varnishes.	04
	(c)	Explain various thermal processes of metals and alloys.	07
		OR	
Q.4	(a)	Write short note on Ceramics.	03
	(b)	Explain Nano materials and their applications.	04
	(c)	Explain manufacturing of Cements.	07
Q.5	(a)	What is the purpose of annealing?	03
	(b)	Write short note on - Dry Corrosion with classification and suitable	04
		diagram.	
	(c)	What is Hardening? Explain the requirements of hardening.	07
0.5	()	OR	0.2
Q.5	(a)	Define: Thermosetting plastics, Composites and Epoxy Resins.	03
	(b)	Mention the properties and applications of PVC.	04
	(c)	How can the corrosion be prevented? Explain various methods.	07