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B.Pharm III Year II Semester (R15) Regular Examinations May/June 2018

PHARMACEUTICAL ANALYSIS - II

Time: 3 hours Max. Marks: 70

PART - A

(Compulsory Question)

- 1 Answer the following: $(10 \times 02 = 20 \text{ Marks})$
 - (a) Define Rf value.
 - (b) What is HETP? Write its significance.
 - (c) Define fronting factor.
 - (d) What are the criteria to perform gas chromatographic analysis?
 - (e) Write the applications of DTA.
 - (f) Write the difference between QC and QA.
 - (g) What is normal phase mode in HPLC?
 - (h) Write the advantages of HPLC over GC.
 - (i) What is optical activity?
 - (j) Define octane rule.

PART - B

(Answer all five units, $5 \times 10 = 50 \text{ Marks}$)

UNIT – I

Write the principle and working of ion exchange chromatography.

OR

What is paper chromatography? Write a note on development of paper chromatogram technique.

UNIT -II

Explain the working of GC with the help of a neat, labeled, schematic block diagram and describe its adsorption isotherm.

OR

5 Discuss briefly detectors and injectors used in GC.

UNIT – III

Write basic principle, instrumentation and applications of DSC in analysis of pharmaceuiticals.

OR

What is validation? Give a brief account of validation of analytical methods.

[UNIT – IV]

8 What is HPLC? Classify them with example. Write its applications in pharmacy.

OR

9 Discuss various parameters in chromatogram in HPLC.

[UNIT – V]

How are x-rays generated? What are the applications of x-ray diffraction? Write a note on Bragg's law.

OR

11 What is RIA? Write its principle and types with example.
