

Code No: B1107/R10

R10**I B.Pharmacy I Semester Supplementary Examinations, Feb/Mar 2014**
PHYSICAL PHARMACY-I**Time: 3 hours****Max Marks: 75****Answer any FIVE Questions**
All Questions carry equal marks

1. Explain the process of vaporization of liquids. Describe the two & phase diagram. [15]
2. (a) What are Miller indices? Illustrate their use and applications.
(b) Explain the term 'pseudomorphism' with suitable examples. [8+7]
3. Explain and derive an expression for the maximum work done when an ideal gas expands isothermally and reversibly. [15]
4. The heat of neutralization of NaOH by HNO₃ is -57.33 MJ/kmol. Calculate the amount of heat liberated when 500cm³ of 1.0N NaOH solution exactly neutralized. [15]
5. (a) Define and write the pharmaceutical significance of Optical rotation and Optical rotatory dispersion.
(b) Define refractive index and mention its significance in pharmacy. [8+7]
6. (a) Explain the method of determination of Optical activity.
(b) Write a note on Induced Dipole moment. [8+7]
7. (a) Write a note on Dalton's law.
(b) Explain how Lowering of Vapour pressure is measured? [7+8]
8. (a) Explain Beckmann's method for the determination of Freezing point depression.
(b) What are the causes for negative and positive deviation from Raoult's law for solutions? Explain with suitable examples. [7+8]
