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Code No: R13104 (R13) (SET - 1)

I B. Tech I Semester Supplementary Examinations, May. - 2018 ENGINEERING CHEMISTRY

(Com. to CE,ME,CSE,PCE,IT,Chem E,Aero E,AME,Min E,PE,Metal E,Textile Engg)
Time: 3 hours

Max. Marks: 70

Note: 1. Question Paper consists of two parts (Part-A and Part-B)

2. Answer **ALL** the question in **Part-A**

3. Answer any **THREE** Questions from **Part-B** PART -A 1. a) Define reverse osmosis. Write its advantages. (3M)b) Differentiate electrolytic cell and electrochemical cell. (4M)c) Explain the role of purity in corrosion. (3M)d) How does the presence of bulky groups on polymer chains affect the properties of (3M)polymer? What is the reason for the knocking in engines? (4M) Explain how laser ablation method is used for manufacture of carbon nanotubes. (5M)PART -B Explain ion exchange process for softening of hard water. (8M)Write the applications of Kohlaursch law. (8M)Explain how nature of metal oxide play important role in oxidation corrosion 3. a) (8M)process. Explain stereoregular polymers and their significance. (8M)Write the advantages and disadvantages of gaseous fuels. (8M)Write the engineering applications of biodegradable polymers and conducting b) (8M)polymers. a) Explain the working of photovoltaic cell and solar reflectors. (8M)b) Explain tinning and electroless plating. (8M)6. a) Explain the working of H_2 - O_2 fuel cell and its applications. (8M)b) Explain ultimate analysis of coal. (8M)7. a) Discuss the properties, preparation and applications of PVC and Styrene (8M)Butadiene Rubber. What is potable water? How can the water purified by sterilization and (8M)disinfection process?