

Roll No.

--	--	--	--	--	--	--	--	--	--	--	--	--	--	--	--	--	--	--	--	--

Total No. of Pages : 02

Total No. of Questions : 10

B.Pharma (2011 to 2016) (Sem.-8)
PHARMACEUTICAL ANALYSIS – III
Subject Code : BPHM-802
Paper ID : [72297]

Time : 3 Hrs.

Max. Marks : 80

INSTRUCTIONS TO CANDIDATES :

1. **SECTION-A** is **COMPULSORY** consisting of **FIFTEEN** questions carrying **TWO** marks each.
2. **SECTION-B** contains **FIVE** questions carrying **FIVE** marks each and students have to attempt any **FOUR** questions.
3. **SECTION-C** contains **FOUR** questions carrying **TEN** marks each and students have to attempt any **THREE** questions.

SECTION-A**Q1 Answer briefly :**

- a) What is red shift in UV?
- b) What is diamagnetic anisotropy?
- c) Explain the terms fluorescence and phosphorescence.
- d) Explain circularly polarized light.
- e) Write down the significance of finger print region in IR.
- f) Why symmetric stretch of CO₂ molecule is inactive in IR?
- g) Define base peak and molecular ion peak in MS.
- h) What is the use of TMS in NMR spectroscopy?
- i) Explain the term optical rotatory dispersion.
- j) Write down the basic principle of polarimetry.
- k) What is mull method in IR?
- l) Why CDCl₃ is used as a solvent in NMR spectroscopy instead of CHCl₃?

- m) Is flame photometry emission spectroscopy? Comment.
- n) Write two pharmaceutical applications of AAS.
- o) Explain the term space lattice.

SECTION-B

- Q2 What is an electromagnetic spectrum? Name the various spectroscopic methods based on electromagnetic radiations.
- Q3 Derive Beer-Lambert law for quantification of compounds by UV spectroscopy.
- Q4 Discuss the applications of X-ray spectroscopy in pharmaceutical analysis.
- Q5 Describe working of a Quadrupole Mass analyser.
- Q6 Explain Spin-spin coupling in ^1H NMR with appropriate examples.

SECTION-C

- Q7 Write an account of diffraction of X-ray by crystals.
- Q8 Explain the working principle of flame photometer by drawing its block diagram. Give quantitative pharmaceutical applications of flame photometry.
- Q9 Explain the following in MS :
 - a) McLafferty rearrangement
 - b) Tropylium ion
 - c) Isotope peaks
- Q10 What are the various instruments used in infrared spectroscopy? Why a double beam spectrophotometer is preferred? Draw a schematic diagram of such instrument. Explain its working describing functioning of individual parts.