Roll No.							Total No.	of	Pages	: 02
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Total No. of Questions: 09

# B.Tech. (BT) (2011 Onwards) (Sem.-5) CHEMICAL ENGINEERING PRINCIPLES

Subject Code: BTBT-501 Paper ID: [A2073]

Time: 3 Hrs. Max. Marks: 60

#### **INSTRUCTIONS TO CANDIDATES:**

- 1. SECTION-A is COMPULSORY consisting of TEN questions carrying TWO marks each.
- 2. SECTION-B contains FIVE questions carrying FIVE marks each and students have to attempt any FOUR questions.
- 3. SECTION-C contains THREE questions carrying TEN marks each and students have to attempt any TWO questions.

## **SECTION-A**

# Q.1 Answer briefly:

- (a) Express a pressure of  $200 \times 10^3$  Pa in mm Hg.
- (b) Define Raoult's law.
- (c) What is Molecularity of the reaction?
- (d) A reaction has a stoichiometric equation  $A + B \rightarrow 2R$ , state the order of the reaction.
- (e) Define space velocity.
- (f) Write the uses of following measurement (i) pH (ii) Viscosity
- (g) Derive the flow equation for orifice meter.
- (h) What are the advantages of cascade control?
- (i) What are the various components of a control system?
- (j) Differentiate closed loop and open loop control.



### **SECTION-B**

Q.2 A reaction was carried out in a batch reactor and the results are reported below. Calculate the rate of the reaction.

Time, t (min)	0	10.0	30.0
% conversion	19.8	46.7	74.0

- Q.3 List various advantages and disadvantages of :
  - (a) Feed forward control system
  - (b) Feed Backward control system.
- Q.4 The heat capacity of ammonia, defined as the amount of the heat required to raise the temperature of a unit mass of ammonia by precisely 1 °C at a constant pressure, is, over a limited temperature range, given by the expression

$$C_P\left(\frac{Btu}{lb_m{}^{\circ}F}\right) = 0.487 = 2.297x10^{-4}T({}^{\circ}F).$$

Determine the expression for  $C_P$  in  $J/(g.^{\circ}C)$  in terms of  $T(^{\circ}C)$ .

- Q.5 What are the factors affecting dynamic response of mercury thermometer?
- Q.6 What the various advantages of head flow-meters over the other flow meters?

## **SECTION-C**

- Q.7 (a) Draw the block diagram of an instrument and explain the function of different functional elements.
  - (b) What is meant by calibration of an instrument and how it is performed?
- Q.8 Define Semi-batch reactors. And state the advantages and disadvantages of batch reactor.
- Q.9 Two methanol water mixtures are contained in separate flasks. The first mixture contains 40wt. % methanol, and second contains 70wt. % methanol. If 200 g of the first mixture combined with 150 g of the second, what are the mass and composition of the product?

**2** M-70502 (S2)-24