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## M.Tech.(Ev.S & E) (Sem.-2) POLLUTION MONITORING TECHNIQUES Subject Code : ES-507 Paper ID : [E0993]

Time: 3 Hrs.

Max. Marks: 100

## **INSTRUCTION TO CANDIDATES :**

- 1. Attempt any FIVE questions out of EIGHT questions.
- 2. Each question carries TWENTY marks.
- 1. Define and contrast between (a) Molarity (b) Molality (c) Normality (d) per cent weight in respect of solutions.
- 2. What are indicator organisms? What are the qualities of indicator organisms? Explain the suitability of Coliform group as indicator organisms. Establish a relationship between pathogens and indicator organisms.
- 3. a) A 100 mL sample containing chloride is titrated with AgNO<sub>3</sub> in a precipitation reaction. Calculate the concentration on Chloride ion in mg/L, if 10 mL of 0.01 N AgNO<sub>3</sub> is used to reach the equivalence point. Discuss the reliability of the analysis.
  - b) An analysis of a sample of water whose pH was measured as 7.6 yielded the following concentrations in mg/L :

N.Y.	Catio	ons	<u>Anions</u>	
L'	Ca <sup>2+</sup>	80	Cl	100
	$Mg^{2+}$	30	$SO_4^{2-}$	201
	Na <sup>+</sup>	72	HCO <sub>3</sub> -	150
	$K^+$	06		

Find the Total hardness, the carbonate hardness, the noncarbonate hardness, and the total alkalinity expressed as  $CaCO_3$ . Estimate the total dissolved solids (TDS) in mg/L. Check the correctness of the chemical analysis. If the pH were 10, what change would you expect in the calculated value of alkalinity? (Take, Ca - 40.1, Mg - 24.3, Na - 23, K - 39.1, Cl - 35.5, S - 32, O - 16, H - 1, C - 12).

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- 4. Discuss the suitability of the following for environmental analysis by taking a suitable example :
  - a) Gravimetric and precipitation methods.
  - b) Chromatography and polarography.
- 5. State, derive and explain the use of the following laws in environmental monitoring
  - a) Lambert's law
  - b) Beer's Law
- 6. a) Draw the schematic diagram of a spectrophotometer and explain the calibration and use. Indicate the suitability of use of UV and VIS spectrometers.
  - b) How does atomic absorption spectrophotometry (ASS) differ from simple spectrometry? Indicate the suitability of AAS.
- 7. a) Draw the schematic diagram of a gas chromatograph. Explain the working.
  - y. www.fitstRanket. b) Discuss the various thermal analysis techniques. Indicate their suitability.

## 8. Write short notes on :

- a) Radioactivity measurement.
- b) NMR.
- c) Mass spectrometry.
- d) XRD.