

Roll No.							Total No. of Pages	s: 02
							i otal itol ol i agos i	

Total No. of Questions: 19

M.Sc (Chemistry)PIT (2015 to 2017) (Sem.-4)

FUNCTIONAL MATERIALS Subject Code: CHL-512A

Paper ID : [74898]

Time: 3 Hrs. Max. Marks: 70

INSTRUCTIONS TO CANDIDATES:

- 1. SECTION-A is COMPULSORY consisting of TEN questions carrying TWO marks each.
- 2. SECTION-B contains SIX questions carrying FIVE marks each and students have to attempt ALL questions.
- 3. SECTION-C contains THREE questions carrying TEN marks each and students have to attempt any TWO questions.

SECTION-A

- 1. What is a liquid crystal phase and how does it differ from solid crystal and liquid amorphous phases?
- 2. What is a "blue phase"?
- 3. Give two examples of liquid crystals in electronics.
- 4. What is superconducting perovskite?
- 5. Define "Fullerene".
- 6. What are nanotubes?
- 7. Why is Graphene almost transparent?
- 8. Give two examples of zeolites.
- 9. Give two examples of photocromic materials.
- 10. What are thermocromic materials? Give two examples and their applications.

1 M-74898 (S39)-2404



SECTION-B

- 11. Write a note on graphene production.
- 12. Give an account of photocromic materials.
- 13. Briefly describe the synthesis of zeolites.
- 14. Write a note on thermal properties of grapheme.
- 15. Explain the structure of fullernes and discuss the unique advantages they possess.
- 16. Write briefly about materials used in light emitting diodes.

SECTION-C

- 17. What are "Mesogens"? What are the requirements of mesogens? Discuss the mesophases of mesogens.
- 18. Classify Perovskites. Discuss their applications giving suitable examples.
- 19. What are Zeolites? Giving suitable examples, discuss their applications in the industry.

2 | M-74898 (S39)-2404