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Rajiv Gandhi University of Health Sciences, Karnataka

II Year B.Pharm Degree Examination - Sep 2012

Time: Three Hours Max. Marks: 70 Marks

PHYSICAL PHARMACEUTICS (Revised Scheme 3)

Q.P. CODE: 2606

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- State and explain Raoults law. What are ideal and real solutions give examples
- 2. What do you mean by rate of reaction? Explain various factors affecting rate of reaction
- 3. Define and classify colloids. List out the methods for preparation of colloids and explain any two methods

SHORT ESSAYS (Answer any Six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. What are clatharates? Give its application with examples
- 5. Define emulsifying agents. On basis of mechanism of action how emulsifying agents are classified
- 6. Write a note on evaluation of stability of suspensions
- 7. How do you represent particle size distribution graphically?
- 8. Explain Brookfield viscometer
- 9. Define dissolution. Discuss the factors affecting dissolution
- 10. Explain Gibb's adsorption equation
- 11. What is pH? Explain Sorensen's pH scale

SHORT ANSWERS

 $10 \times 2 = 20 \text{ Marks}$

- 12. What is Snell's law?
- 13. List out various physical properties of drug molecule
- 14. Give any two applications of distribution law
- 15. Give Arrhenius equation
- 16. Define HLB value
- 17. Give the applications of porosity
- 18. Write a note on micro emulsions
- 19. Classify complexes
- 20. What is Faraday Tyndall effect?
- 21. Give buffer equation