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Rajiv Gandhi University of Health Sciences, Karnataka

II Year B.Pharm Degree Examination – Mar 2013

Time: Three Hours Max. Marks: 80 Marks

PHYSICAL PHARMACEUTICS

(Revised Scheme - 2)

Q.P. CODE: 1956

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. List the different methods available for determination of order of a reaction. Explain any one method for the determination of the order of a reaction
- 2. Discuss the Rheograms for Newtonian and dilatant materials. Give the examples each for the above systems
- 3. What are colloids? Explain as to how viscosity property can be made use of to establish the Molecular weight of a collidal dispersion

SHORT ESSAYS (Answer any Eight)

 $8 \times 5 = 40 \text{ Marks}$

- 4. Discuss the steps involved in the preparation of a pharmaceutical buffer
- 5. Define dipole moment. How dipole moment can be correlated with biological activity in case of DDT
- 6. Explain as to how principle of partition Co-efficient are useful in case of multiple extraction in contrast to a single extraction
- 7. Discuss Harkin's oriented wedge theory to account for the type of emulsion formed
- 8. Explain as to what do you mean by steady state diffusion
- 9. What are sedimentation parameters? Explain
- 10. Discuss the merits and demerits of optical microscopy for particle size determination
- 11. Deduce freundlich adsorption isotherm and give its graphical representation
- 12. What is spreading coefficient? Give its significance
- 13. Write a note on monomolecular inclusion complexes and their applications

SHORT ANSWERS 10 x 2 = 20 Marks

- 14. Differentiate between a chelate and an inorganic metal complex with the help of suitable examples
- 15. Define phase volume ratio and phase inversion in emulsions
- 16. By what mechanism cholesterol and lecithin act as emulsifuing agents
- 17. Discuss the concept of any two equivalent spherical diameters
- 18. Define critical micelle concentration
- 19. Give two examples for hydrophobic colloids
- 20. What is a Rheopexy
- 21. What is critical solution temperature
- 22. What are microemulsions
- 23. Define electrophoresis and electro-osmosis



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