

# Rajiv Gandhi University of Health Sciences, Karnataka I Year B.Pharm Degree Examination - Dec-2014

Time: Three Hours Max. Marks: 70 Marks

## PHARMACEUTICAL ORGANIC CHEMISTRY – I (Revised Scheme 3)

**Q.P. CODE: 2604** 

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

#### LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$ 

- 1. Explain the stability of conjugated dienes and mechanisms of 1,2 and 1,4 additions with examples.
- 2. Discuss the mechanisms of
  - i) Halogenation of benzene
  - ii) Dehydration of alcohols
- 3. Write the Kinetics, mechanism, stereochemistry and reactivity of SN₂ reaction.

### **SHORT ESSAYS (Answer any Six)**

 $6 \times 5 = 30 \text{ Marks}$ 

- 4. Define the following
  - a) Electrophile
  - b) Nucleophile
  - c) Isomerism
  - d) Free radical
  - e) Carbocation
- 5. Write the mechanism involved in chlorination of methane.
- 6. Write a note on Markovnikov's rule.
- 7. Explain Hyper conjugation.
- 8. Discuss Benzoin condensation with mechanism.
- 9. Explain Sandmeyer's Diazocoupling reactions.
- 10. Explain acidity of carboxylic acids.
- 11. Explain Bayer's strain theory.

#### **SHORT ANSWERS**

 $10 \times 2 = 20 \text{ Marks}$ 

- 12. Write the IUPAC names of the following a) acetone b) Glycerol
- 13. Write the reaction of Frie's rearrangement.
- 14. Write the reaction of Friedel Crafts Alkylation.
- 15. Compare the basicity of: methyl amine and aniline.
- 16. How do you convert: carboxylic acid to ester?
- 17. Write the structures of
  - a) 2-pentenal
  - b) 2,4 dimethyl pentane
- 18. Write the classification of carbocations with examples.
- 19. What is Knoevenagel condensation?
- 20. Define homolysis and heterolysis with examples.
- 21. Define hydrogen bonding. Classify with examples.

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