Code No: R10204/R10

Set No. 1

I B.Tech II Semester Supplementary Examinations, February 2013 ENGINEERING CHEMISTRY -II

(Common to Civil Engineering, Electrical & Electronics Engineering, Mechanical Engineering, Electronics & Communication Engineering, Computer Science & Engineering, Chemical Engineering, Electronics & Instrumentation Engineering, Bio-Medical Engineering, Information Technology, Electronics & Computer Engineering, Aeronautical Engineering, Bio-Technology, Automobile Engineering, Mining and Petroliem Technology)

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. (a)Give the differences' between Chain polymerization & Step growth polymerization?
 - (b) Write a brief account of emulsion polymerization
 - (c) Explain the interpretation of conductivity in conjugated polymers? [5+5+5]
- 2. (a)Distinguish between Thermo settings & Thermo plastics?
 - (b) Explain with neat sketchs any two methods of moulding of plastics? [6+9]
- 3. (a) What is an elastomer. Explain the characteristic of clastomers
 - (b) Write short note on Gutta Percha.
 - (c) Differentiate between a natural rubber and an elastomer. [7+3+5]
- 4. (a) How various types of carbon nano tubes can be formed from grapheme?
 - (b) Discuss how nano technology usefull.
 - (c) What are the advancements of nano technology in Electronics Field [7+4+4]
- 5. (a) Discuss the merits and demerits of dry and wet process for the manufacture of Portland cement
 - (b) What are ceramics? How are ceramics glazed?
 - (c) Write any two types of refractories and their uses [5+5+5]
- 6. (a) Write a short note on theories, which have been put forward to explain the origin of petroleum.
 - (b) Discuss any four important properties of a good lubricant [8+7]
- 7. Define corrosion of metals. What are different types of corrosion? Explain the electrochemical theory of wet corrosion giving its mechanism. [8+7]
- 8. Green environment is a safer environment. What measures have to be taken for this and explain them in detail with suitable examples [15]

Code No: R10204/R10

Set No. 2

I B.Tech II Semester Supplementary Examinations, February 2013 ENGINEERING CHEMISTRY -II

(Common to Civil Engineering, Electrical & Electronics Engineering, Mechanical Engineering, Electronics & Communication Engineering, Computer Science & Engineering, Chemical Engineering, Electronics & Instrumentation Engineering, Bio-Medical Engineering, Information Technology, Electronics & Computer Engineering, Aeronautical Engineering, Bio-Technology, Automobile Engineering, Mining and Petroliem Technology)

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. (a)Write an account of the preparation, properties & engineering applications of the following. (i)PVC (ii)Teflon
 - (b) Write a brief account of following. (i) Tacticity of polymer (ii) Functionality of polymer [9+6]
- 2. (a)Distinguish between Thermo settings & Thermo plastics?
 - (b) Explain with neat sketchs any two methods of moulding of plastics? [6+9]
- 3. (a) What are the draw backs of Natural Rubber?
 - (b) What is compounding of Rubber; discuss one suitable method for compounding rubber [6+9]
- 4. (a) Explain SWNT & MWNT
 - (b) Describe any one method for the production of nanotubes.
 - (c) Descuss the application of fullerenes

[5+7+3]

- 5. (a) Write the chemical reactions that take place (along with temperature) during the manufacture of portland cement
 - (b) What are ceramics? Discuss their classification

[8+7]

- 6. (a) Write a short note on theories, which have been put forward to explain the origin of petroleum.
 - (b) Discuss any four important properties of a good lubricant

[8+7]

- 7. (a) What is the basic difference between a paint and varnish.
 - (b) What are the properties and functions of the constituents of a paint [8+7]
- 8. (a) What is Green Chemistry? Write briefly about Engineering Applications of Green Chemistry?
 - (b) Discuss any four Principals of the Green Chemistry.

[7+8]

Code No: R10204/R10

Set No. 3

I B.Tech II Semester Supplementary Examinations, February 2013 ENGINEERING CHEMISTRY -II

(Common to Civil Engineering, Electrical & Electronics Engineering, Mechanical Engineering, Electronics & Communication Engineering, Computer Science & Engineering, Chemical Engineering, Electronics & Instrumentation Engineering, Bio-Medical Engineering, Information Technology, Electronics & Computer Engineering, Aeronautical Engineering, Bio-Technology, Automobile Engineering, Mining and Petroliem Technology)

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. (a) How are tensile strength and optical properties influenced by the structure of plastics
 - (b) What is meant by Coordination polymerization? Explain its mechanism? [9+6]
- 2. (a)Distinguish between Thermo settings & Thermo plastics?
 - (b) Explain with neat sketchs any two methods of moulding of plastics? [6+9]
- 3. What are the additives mixed with natural rubber to improve required proprieties?

 Disuses about the different additives briefly?

 [15]
- 4. (a)Describe the production of Carbon Nano Tubes by CVD method
 - (b) Explain the structure of C_{60} .
 - (c) Write notes on Quantum dots

[8+3+4]

- 5. (a) Explain how Portland cement is manufactured?
 - (b) What are refractories? Why refractoriness under load is an important property of a refractory? [8+7]
- 6. (a) What is meant by knocking? How it is related to chemical constitution?
 - (b) Define and Signify (i) Flash point and fire point (ii) Could & Pour point [7+8]
- 7. Define Anodic and cathodic protection and explain the sacrificial anodic protection and impressed current cathodic protection. [15]
- 8. Green environment is a safer environment. What measures have to be taken for this and explain them in detail with suitable examples [15]

Code No: R10204/R10

Set No. 4

I B.Tech II Semester Supplementary Examinations, February 2013 ENGINEERING CHEMISTRY -II

(Common to Civil Engineering, Electrical & Electronics Engineering, Mechanical Engineering, Electronics & Communication Engineering, Computer Science & Engineering, Chemical Engineering, Electronics & Instrumentation Engineering, Bio-Medical Engineering, Information Technology, Electronics & Computer Engineering, Aeronautical Engineering, Bio-Technology, Automobile Engineering, Mining and Petroliem Technology)

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. (a) Explain the term polymerization and co-polymerization with suitable examples? (b) What are the conducting polymers, name four of them & give their engineering
 - applications? [9+6]
- 2. (a) What is meant by the moulding? Explain with neat diagram compression & injection moulding of plastics?
 - (b) Write a note on engineering applications of the plastics? [8+7]
- 3. (a) Write notes on preparation, properties and uses of styrene-Butadiene Rubber (b) What are the charastricts of polyurethane, Buna-N, Neoprene and Butyl rubber
- 4. (a)Describe the production of carbon nano tubes by chemical vapour deposition
 - (b) How fullerenes are produced

[7+8]

(c) write briefly about the carbon nano tubes

[7+4+4]

- 5. (a) What are the constituents of cement?
 - (b) Discuss about the classification of ceramics
 - (c) Write short notes on properties of refractories

[5+5+5]

- 6. (a) What are anti knocking agents? Describe the functioning of TEL.
 - (b) Write short notes on fractional distillation of petroleum.
 - (c) How are lubricants classified?

[5+5+5]

- 7. Define Anodic and cathodic protection and explain the sacrificial anodic protection and impressed current cathodic protection. [15]
- 8. What is green chemistry? Explain the phase transfer catalyst for green synthesis, ultra sound assisted method for green synthesis and application of green chemistry. [15]