R09

Set No. 2

II B.Tech II Semester Examinations, APRIL 2011 BIOPROCESS ENGINEERING Bio-Technology

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

1. Write short notes on:

Code No: R09222305

- (a) Maximum Biomass Yield.
- (b) Product stoichiometry.

[7+8]

- 2. Give summary of some of the important parameters, Phenomena and interactions which determine cell population kinetics. What are primary and secondary metabolites with suitable examples? [15]
- 3. Compare and contrast a batch bioreactor and continuous bioreactor. Mention the relative advantages and disadvantages of each. [15]
- 4. Write short notes on the following:
 - (a) inhibitors added to media
 - (b) restricted nutrient levels in media

[7+8]

- 5. What do you mean by upstream and downstream process? Explain various unit operations involved in them. [15]
- 6. Determine standard heats of reaction from heats of combustion. [15]
- 7. Steam output of boiler is measured by measuring feed water. The tank level reading from 8.00 a.m. to 8.00 p.m. was 600 m³. Continuous blow down was given at 1% of the boiler feed rate during the above period. Find out the average actual steam delivered per hour? [15]
- 8. The growth of bakers yeast (S. cerevisiae) on glucose may be simply described by the following equation:

 $C_6H_{12}O_6 + 3O_2 + 0.48NH_3 \rightarrow 0.48C_6H_{10}NO_3 + 4.32H_2O + 3.12CO_2$ In a batch reactor of volume 10 l, the final desired concentration is 50gdw/l. Using

the above reaction stoichiometry:

- (a) Determine the concentration and total amount of glucose in the nutrient medium
- (b) Determine the yield coefficients $\mathbf{Y}_{X/S}$ (biomass/glucose) and $\mathbf{Y}_{X/O2}$ (biomass/oxygen)

|15|

Code No: R09222305

R09

Set No. 4

II B.Tech II Semester Examinations, APRIL 2011 BIOPROCESS ENGINEERING Bio-Technology

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

1. Write about heat evaluation in aerobic cultures.

2. Write short notes on:

(a) Total balance
(b) Component balances.

[7+8]

3. Discuss in detail the requirement for growth and formulation of media.

[15]

4. Write short notes on:

(a) Degree of reduction
(b) Elemental balances.

[7+8]

5. Write about simple unstructured model for microbial growth.

[15]

6. Outline the various steps involved in integrated Bioprocess. [15]

7. Fumeric acid is produced from mallic acid using the enzyme, fumerase. Calculate the standard heat of reaction for the following enzyme transformation;
 C₄H₆O₅ → C₄H₄O₄ + H₂O
 (Mallic acid → Fumeric acid + water)

8. Write a brief note on biosensors, and describe how they are used for measuring process parameters. [15]

R09

Set No. 1

II B.Tech II Semester Examinations, APRIL 2011 BIOPROCESS ENGINEERING Bio-Technology

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. Why should we do energetic analysis? What is Gibbs energy balance for a growing cell. [15]
- 2. What is bioreactor and write about the basic functions of a bioreactor for microbial or animal cell culture? [15]
- 3. Write shorts note on:

Code No: R09222305

- (a) Reference states
- (b) State properties [7+8]
- 4. Write Material balance equations for unsteady state. [15]
- 5. Write in detail about growth of filamentous organisms. [15]
- 6. What factors do you consider as essential for a successful design of fermenter and the various parameters that need to be controlled for successful operation of fermenter.

 [15]
- 7. Aerobic degradation of benzoic acid by a mixed culture of microorganisms can be represented by the following reaction $C_6H_5COOH(substrate) + aO_2 + bNH_3 \rightarrow cC_5H_7NO_2(bacteria) + dH_2O + eCO_2$ Determine the coefficients a, b, c, d and e where RQ = 0.9. [15]
- 8. How the addition of precursors, inhibitors and inducers to the medium help regulate the fermentation process? [15]

Code No: R09222305

R09

Set No. 3

II B.Tech II Semester Examinations, APRIL 2011 BIOPROCESS ENGINEERING Bio-Technology

Time: 3 hours Max Marks: 75

L'im	Answer any FIVE Questions All Questions carry equal marks *****	KS: 75
1.	Explain energy balance equations.	[15]
2.	Write short notes on the following:	
	(a) Factors influencing the choice of nitrogen source.(b) Media formulation.	[7+8]
3.	Explain aerobic and anaerobic metabolism with respect to energetics.	[15]
4.	Explain the following: (a) Basic differences between fermentation and chemical reaction. (b) Basic components of a biotechnological process line.	[8+7]
5.	Assume that experimental measurements for a certain organism have show cells can convert two-thirds (wt/wt) of the substrate carbon (alkane) to bio Calculate the stochiometric coefficients for the following biological reaction: Hexadecane: $C_{16}H_{34} + aO_2 + bNH_3 \rightarrow c(C_{4.4}H_{7.3}N_{0.86}O_{1.2}) + dH_2O + eCO_2$	n that mass.
6.	Explain Material balance equations for recycle, bypass and purge streams.	[15]
7.	Explain substrate and product inhibitions analyses.	[15]
8.	Describe various physical process parameters that need to be controlled in mentation process. Explain them briefly.	a fer- [15]